

# COMPARISON OF THE OXYGEN CONCENTRATION IN CZOCHRALSKI SILICON CRYSTAL OBTAINED BY A SIMPLE LUMPED-PARAMETER MODEL AND SOPHISTICATED 2D-3D SIMULATIONS

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## Motivation

- Numerical modelling is widely used to improve the Czochralski (Cz) crystal growth technology in order to obtain lower oxygen contents in silicon crystals
- Highly precise numerical simulations, e.g. including calculation of turbulent melt flow, typically take several weeks and need high performance computing clusters
- Our approach is to evaluate which information about oxygen concentration in the silicon crystal can be achieved by a simple, time- and cost saving boundary layer model
- The results of this lumped model are compared to that obtained by a highly sophisticated 2D/3D coupled numerical Cz model for 24" crucible hot zone ([1],[2])
- Oxygen content was calculated in dependence on the crucible and crystal rotation for different crystal lengths

## Calculation of O concentration

From the balance of the mass fluxes

$$\frac{dn_m}{dt} = 0 = J_c A_c - J_s A_s - J_x A_x \quad \text{Eq. 4}$$

the pseudo-steady concentration inside the melt volume  $C_m$  can be obtained

$$C_m = \frac{1}{1 + \frac{A_s \delta_c'}{A_c \delta_s'} + \frac{A_x \delta_c'}{A_c D} + \frac{A_x \delta_c'}{A_c D} + \frac{A_x \delta_c'}{A_c D} + \frac{A_x \delta_c'}{A_c D} + \frac{A_x \delta_c'}{A_c D} + \frac{A_x \delta_c'}{A_c D} + \frac{A_x \delta_c'}{A_c D} + \frac{A_x \delta_c'}{A_c D} + \frac{A_x \delta_c'}{A_c D}}{k_{eff}} C_c \quad \text{Eq. 5}$$

The equilibrium oxygen concentration at the crucible wall  $C_c$  can be assumed to be equal to  $C_c = 2 \times 10^{18} \text{ cm}^{-3}$  and therefore independent from the melt overheating  $\Delta T$ .

$k_{eff}$  is the effective segregation coefficient according to Burton Prim Slichter.

The diffusion boundary layers  $\delta_i$  are replaced by the corresponding momentum boundary layers  $\delta_i'$  using the relation

$$\delta_i = 2.4 * Sc^{-1/3} \delta_i' \quad \text{with } Sc = \nu/D \quad \text{Eq. 6}$$

The momentum boundary layers  $\delta_s'$  at the free melt surface depends on the Marangoni number  $Ma$ :

$$\delta_s' = Ma^{-1/3} (r_c - r_x) \quad \text{with } Ma = \frac{-(\frac{\partial \sigma}{\partial T}) \Delta T (r_c - r_x)}{\rho \nu^2} \quad \text{Eq. 7}$$

The momentum boundary layers  $\delta_c'$  at the crucible wall scales with the Ekman number  $Ek$  for the geostrophic flow regime or with the Grashof number  $Gr$  for non rotational flows.

$$\delta_c' = 3Ek^{1/2} r_c \quad \text{with } Ek = \frac{\nu}{2\omega_c r_c^2} \quad \text{Eq. 8}$$

$$\delta_c' = Gr^{-1/4} r_c \quad \text{with } Gr = \frac{\beta g \Delta T r_c^3 (\frac{h}{r_c})}{\nu^2} \quad \text{Eq. 9}$$

For the calculation of the  $Ma$  and  $Gr$  numbers a characteristic temperature difference is needed which is assumed to be equal to the maximum melt over heating  $\Delta T$ . From the 2D/3D simulations for the given 24" hot zone it was found that  $\Delta T$  depends mainly on the crucible rotation rate  $\omega_c$  as follows:

$$\Delta T \approx 27K + 2\omega_c \left[ \frac{K}{rpm} \right] \quad \text{Eq. 10}$$

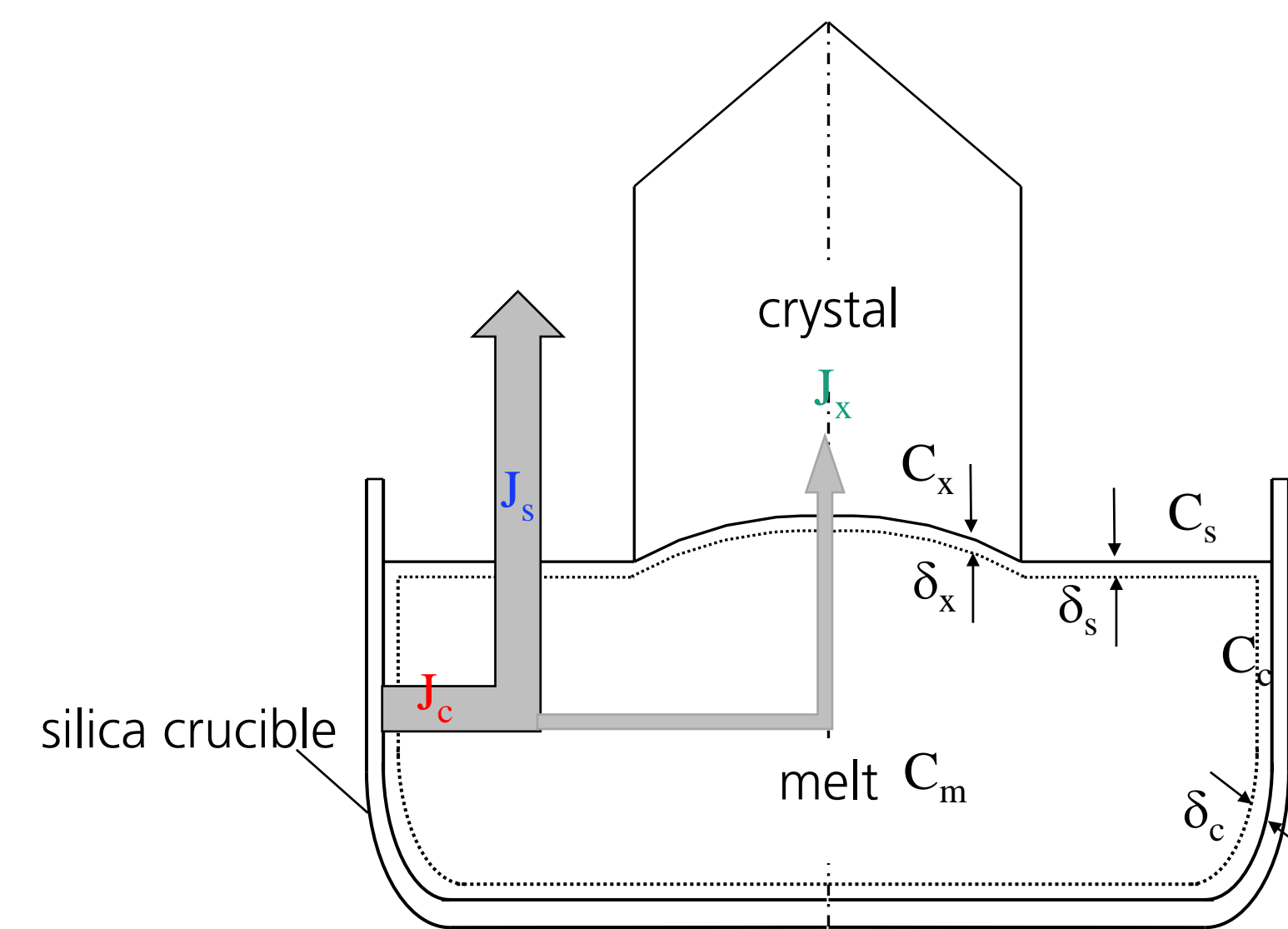
With eqs. 5-10 the oxygen concentration  $C_m$  can be calculated as follows

$$C_m = \frac{1}{1 + \frac{A_s \delta_c'}{A_c \delta_s'} + \frac{A_x \delta_c'}{A_c (D/V)} + \frac{2.4 k_{eff}}{Sc^{1/3}}} 2 \times 10^{18} \text{ [atoms/cm}^3] \quad \text{Eq. 11}$$

## Conclusion

- The presented simple lumped-model can be used for rough estimation of oxygen concentration in Cz-crystals, e.g. in dependence of ingot length or crucible rotation, which make it interesting for beginners in crystal growth
- However, if high precise analysis of oxygen distribution in crystal and melt is required the lumped model hits the wall and the complex and time consuming 2D/3D simulations are indispensable

## The boundary layer model



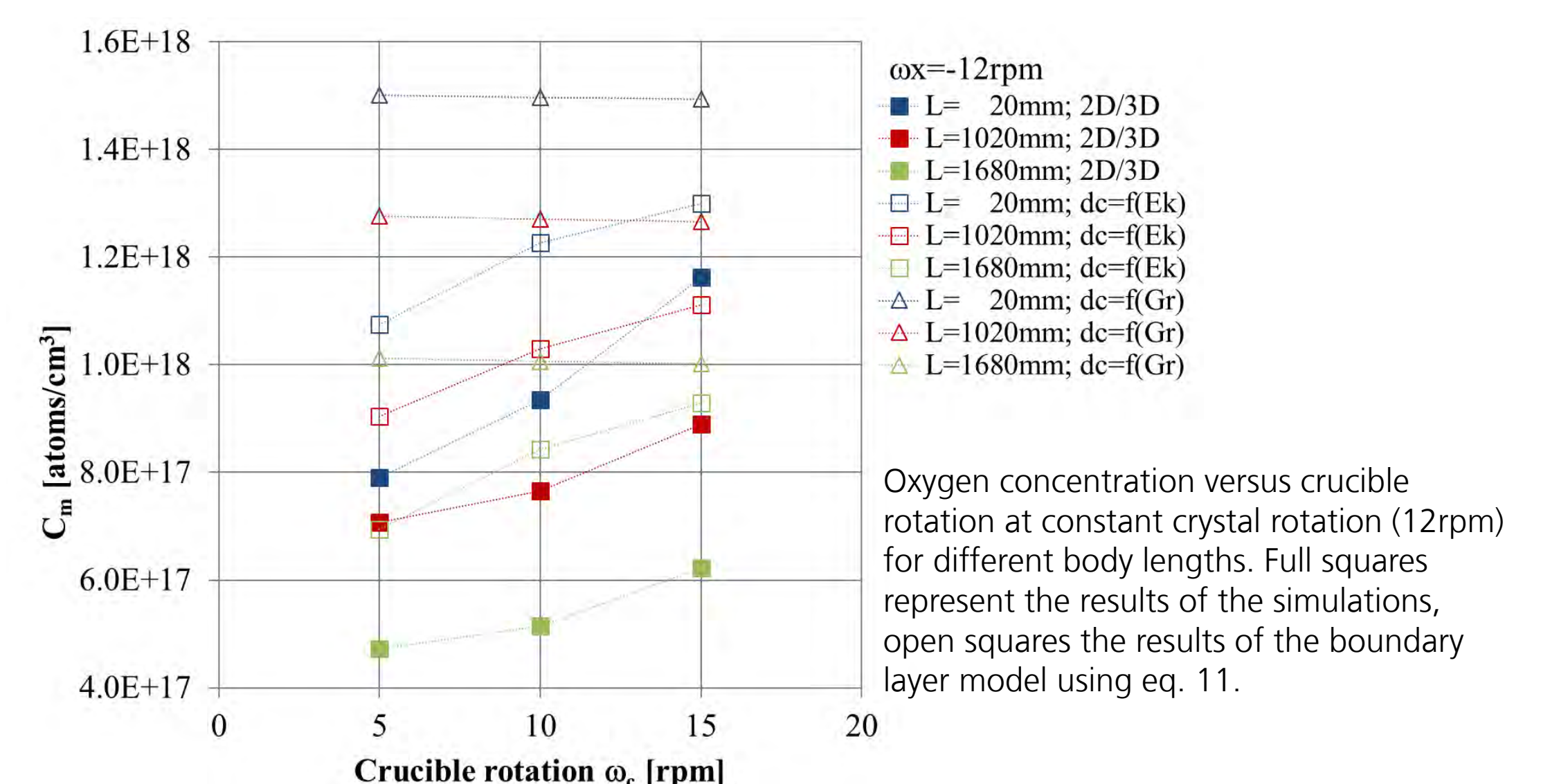
Variable	Description
$C_c$	Equilibrium concentration at crucible wall
$C_m$	Solute concentration in the melt volume
$C_s$	Solute concentration at the free melt surface
$C_i$	Solute concentration at the s/l interface
$A_c$	Contact area between melt and crucible wall
$A_s$	Area of free melt surface
$A_x$	Area of s/l interface
$D$	Diffusion coefficient
$\nu$	Kinematic viscosity
$r_c$	Crucible radius
$r_x$	Crystal radius
$\delta_s$	Solute boundary layer thickness
$n_s$	Molar flow rate
$J$	Molar flux
$\omega$	Angular velocity
$V$	Pull speed

Oxygen mass fluxes

- Source:** Dissolution of the SiO<sub>2</sub> crucible  $J_c A_c$   $\dot{n}_c = J_c A_c$   $J_c = \frac{D}{\delta_c} (C_c - C_m)$  Eq. 1
- Sinks:** Evaporation of SiO at free melt surface  $J_s A_s$   $\dot{n}_s = J_s A_s$   $J_s = \frac{D}{\delta_s} (C_m - C_s)$  Eq. 2
- Incorporation into the crystal  $J_x A_x$   $\dot{n}_x = k_{eff} V C_m A_x$  Eq. 3

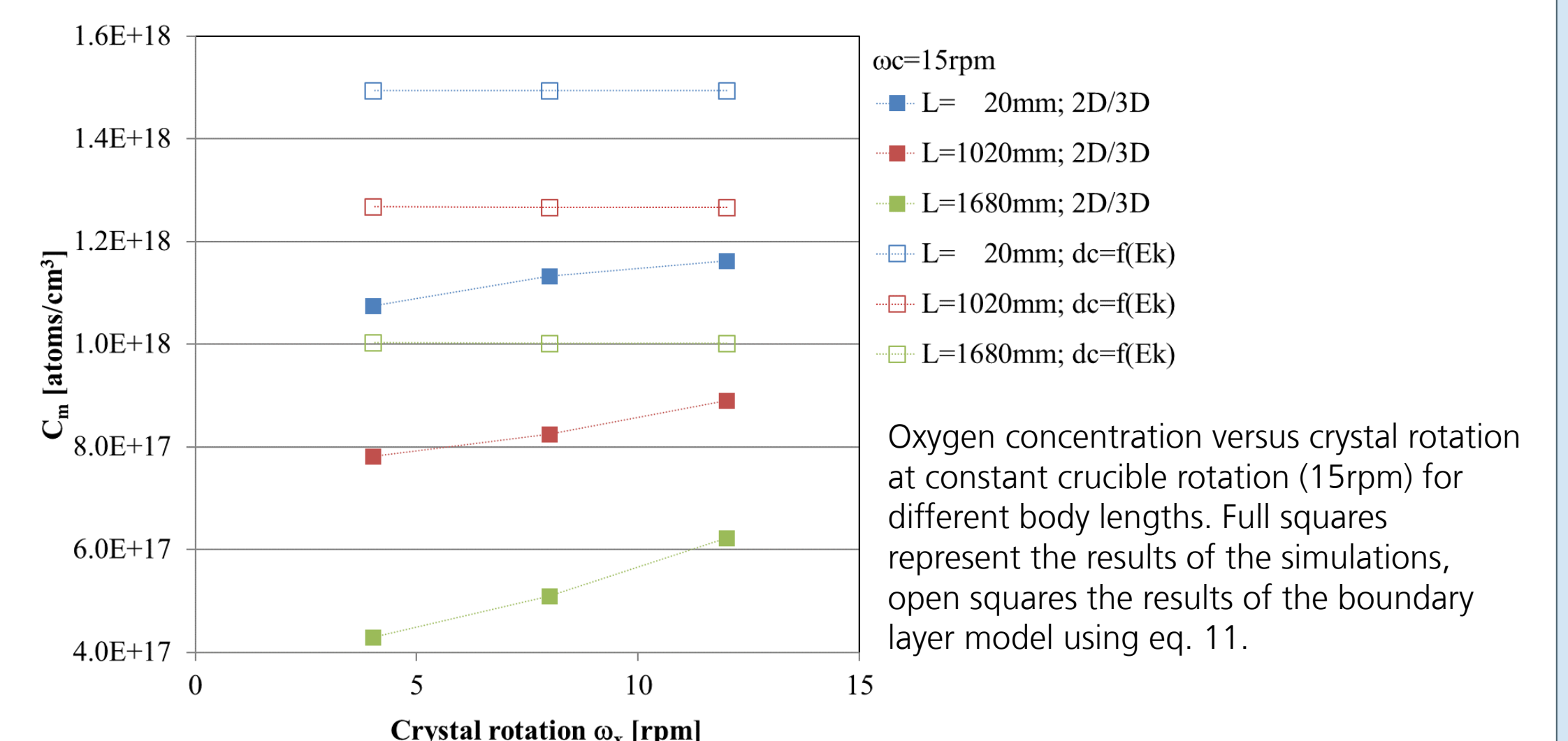
## Comparison to 2D/3D simulations

Oxygen concentration in silicon crystal in dependence on crucible rotation  $\omega_c$  for constant crystal rotation  $\omega_x$



- Qualitative agreement between lumped model and 2D/3D simulations using Ekman number (eq.8) in the lump-model: The increasing oxygen trend becomes visible (decreasing  $\delta_c$  at the crucible wall with increasing  $\omega_c$ , resulting in increasing  $J_c$ )
- No matching to 2D/3D model if  $Gr$  number is used for calculation of  $\delta_c$

Oxygen content in silicon crystal in dependence on crystal rotation  $\omega_x$  for constant crucible rotation  $\omega_c$



- Slight increase of the oxygen concentration with increasing crystal rotation in the 2D-3D simulation is not reproduced by the boundary layer model
- Complex flow structure itself is contributing much to the real oxygen transport

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